

Millimetre-wave spectroscopy of N-cyano metilamine

Luca Bizzocchi, Domenico Prudeniano

*Center for Astrochemical Studies, Max-Planck-Institut für extraterrestrische Physik,
Gießenbachstr. 1 85748 Garching bei München, Germany*

N-cyano metilamine (CH_3NHCN) is a structural isomer of amino acetonitrile ($\text{NH}_2\text{CH}_2\text{CN}$), an important astrophysical molecule thought to be chemically related to (possibly a direct precursor of) the simple amino acid, glycine. Amino acetonitrile has been detected towards the galactic centre in Sgr B2(N). Its fractional abundance was estimated to be 2×10^{-9} , with an excitation temperature, $T_{\text{exc}} \sim 100$ K. Observational evidences point to a chemical pathway based on grain-surface chemistry, and the detection of its isomer with the $-\text{CN}$ moiety on the amine group would provide useful hints to constrain the formation pathway of more complex nitrogen bearing species.

The rotational spectrum of CH_3NHCN is rather complex due to the presence of two quadrupolar ^{14}N nuclei, and two large amplitude motions: the methyl internal rotation and the amine $-\text{H}$ inversion through the ab principal axis plane. While the hyperfine structure is not resolvable in the millimetre regime, the internal dynamics of the molecule produces a significant complexity. Tunnelling between both torsional and inversion potential minima splits each J_{KaKc} rotational levels into four substates labelled A^+ , E^+ , A^- , and E^- . The magnitude of the two effects is similar, thus no simple factorisation of the Hamiltonian into two “simple” internal rotation subproblems (one each inversion species) holds strictly.

N-cyano metilamine is unstable in ordinary laboratory conditions thus the isolation of a pure sample prepared by ordinary organic chemistry is not feasible. In our study, we adopted a pyrolysis methods which starts from stable precursors (e.g., dimethyl cyanamide, $(\text{CH}_3)_2\text{NCN}$). A steady, slow flow of fresh pyrolysis product was then kept into the absorption cell while the spectrum is recorded; this provided the strongest absorption signals, even if the spectral contamination from side products turned out to be a serious issue.

We carried out a preliminary study of the rotational spectrum of CH_3NHCN in selected frequency regions spanning the 80-350 GHz interval. To date, lines belonging to all four substates were identified and analysed through an “effective-fit” or “band-by-band” approach, i.e., employing a different Watson-type Hamiltonian for each rotation-inversion state. This method turned out to moderately successful for the ‘+’ states, but to a much lesser extent for the ‘-’ states.